

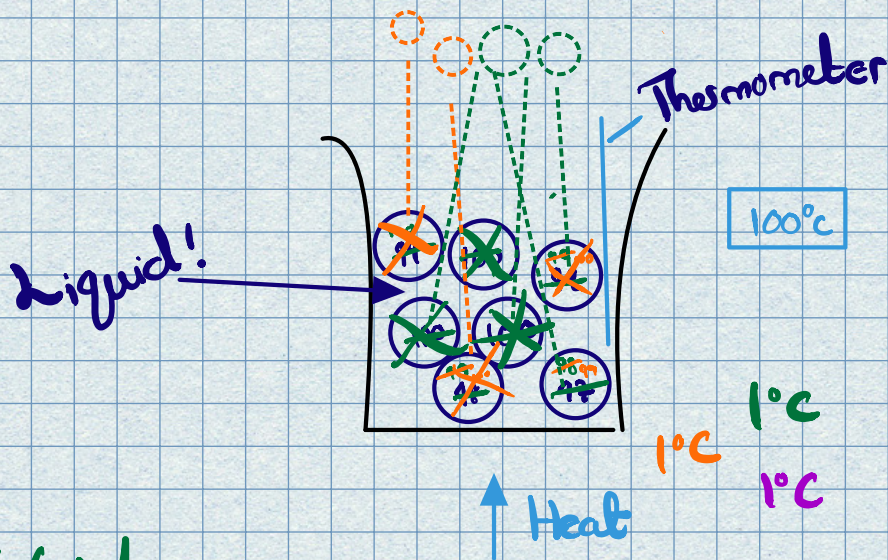
Melting Point :-

A constant temperature when a solid converts into liquid!

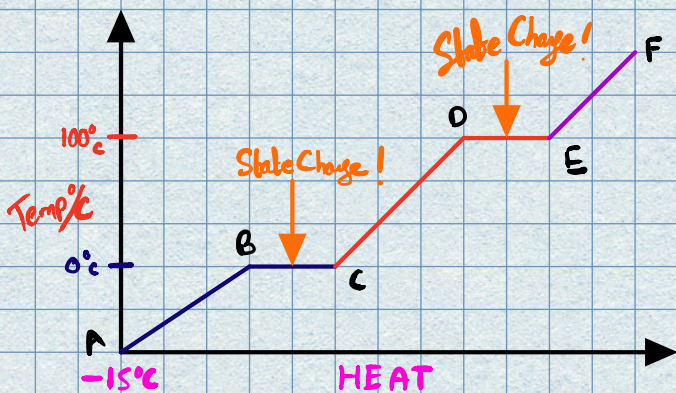
Boiling Point :-

A constant temperature when a liquid converts into gas!

water = B.P = 100°C



Heating Curve of Water :-



A-B = Solid

B-C = Solid + liquid

C-D = liquid

D-E = liquid + Gas

At the time of boiling, temperature remains constant while heat is provided, at **Boiling Point**, heat given is used to convert **liquid** into **Gas** or to break down the intermolecular forces of **liquid** particles

★ Boiling :- ENDO ! ★ Evaporation :- ENDO !

✓ liquid → Gas

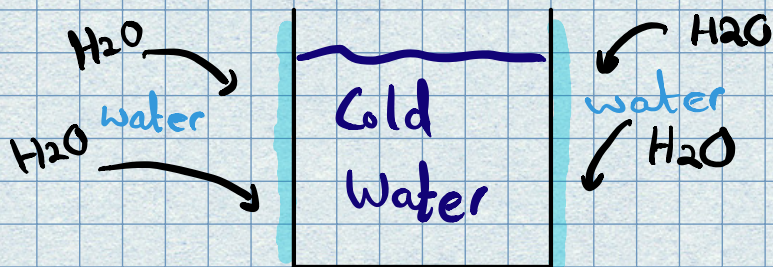
✓ liquid → Gas

→ Fixed temp

→ Any temp

→ Throughout the liquid

→ Only on the surface of liquid !

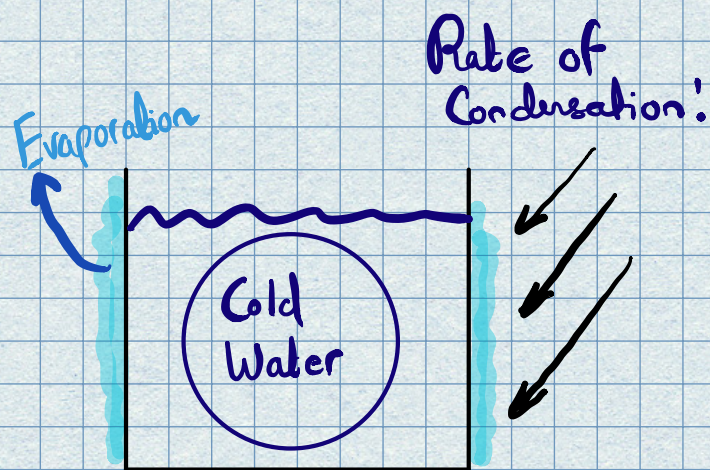


CONDENSATION!

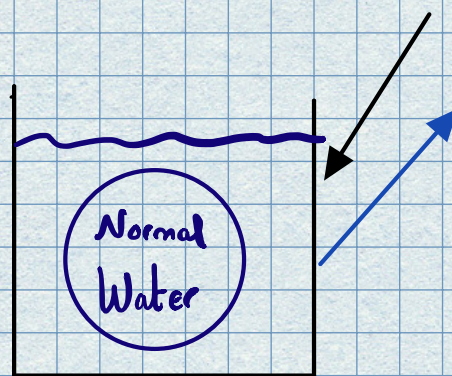


Vapors → liquid !

→ For Condensation of Vapours of a liquid Vapours may condense just below 1°C of its Boiling Point !



Rate of condensation = $R_o E$



Rate of Condensation > Rate of Evaporation

